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2H-1,2,3-Diazaphospholes in Addition Reactions

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2H-1,2,3-DIAZAPHOSPHOLES IN ADDITION REACTIONS

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2H-1,2,3-Diazaphospholes activity depends on the nature of substituents in 2,4,5-positions of the diazaphosphole ring. 5-Methyl-2-phenyl(acetyl)-1,2,3-diazaphospholes $\underline{1}$, $\underline{2}$ react with the alcohols at room temperature, and 2-cyclohexyl-4,5-tetramethylen-1,2,3-diazaphosphole $\underline{3}$ - at 65°C. The adduct $\underline{4}$ degrades to cyclohexanon-N-cyclohexyl-N-methyl-hydrasone 5.

Diazaphosphole $\underline{3}$ is less active than diazaphosphole $\underline{1}$ in the reaction with cyclohexanon. The treatment of diazaphosphole $\underline{3}$ with cyclohexanon leads to the compound $\underline{6}$: the result of the 1:1-hydrolysis adduct. Two routs of the reaction were found in the case of diazaphosphole $\underline{1}$: via cyclohexanon addition to the $\lambda^3\sigma^2P=C$ bond and at the $\lambda^3\sigma^2P$ atom with formation of the product similar to $\underline{6}$. The interaction of diazaphospholes $\underline{1}$, $\underline{2}$, $\underline{3}$ with phenylazide gives trimers 7.

Diazocompounds add to diazaphospholes by 1,3-dipolar cyclo-addition (${\rm AcCHN}_2$, (EtO) $_2{\rm P(O)CHN}_2$, EtOC(O)CHN $_2$, ${\rm Ph}_2{\rm CN}_2$, PhCHN $_2$, diazofluoren to diazaphospholes 1,2). The existence of intermediate diazaphosphole ylides is ascertained in the reactions with Me $_2{\rm CN}_2$, i-PrCHN $_2$, t-BuCHN $_2$. The availability of geometric isomeres of phosphirans is established in the case of MePhCN $_2$.